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(54) COMPOSITION FOR THE AMELIORATION OF MALODOURS

(71) We, WARNER-LAMBERT COM-PANY, of 201 Tabor Road, Morris Plains, New Jersey 07950, United States of America, a corporation organised under the laws of the State of Delaware, United States of America, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

The present invention is concerned with new compositions, such as mouthwashes, mints, breath sprays and the like, having included therein one or more selected compounds.

We have found that the odour-masking qualities of mouthwashes, mints, breath sprays, toothpastes and the like are unexpectedly enhanced by the addition of one or more of a selected group of compounds. These compounds, which we call "reodorants" are terpenes which are distinguished by their ability to enhance the odour-masking efficiency of other compounds, when the former are present in minor amounts.

The odour-masking properties or deodorant properties of flavours and fragrances are well known. Flower oils have well known odour-masking properties but, when used alone, give rise to peculiar and, in many cases, unpleasant sensations. It is, therefore, apparent that not every fragrance or flavour would be aesthetically acceptable in diminishing malodours nor, for that matter, capable of reducing them at all.

The reodorant compounds which we have found to be suitable for use in the compositions of the present invention are α - ionone, α - methyl - ionone, citral, geranyl formate and geranyl acetate, which are terpenes. It has been further determined that only a few terpenes possess reodorant qualities although, in large amounts, many terpenes can be described as flavours and per-

Thus, according to the present invention,

there is provided a composition for the amelioration of malodours, comprising a known deodorant composition with a minor content of at least one compound selected from α - ionone, α - methyl - ionone, citral, geranyl formate and geranyl acetate.

Determination of reodorant properties of various substances has been carried out both in vivo and in vitno. The results obtained using both methods show good correlation in measuring reodorant properties. The method of measuring reodorance in each case was organoleptic, i.e. panels of judges skilled in making such determinations were used to evaluate the effectiveness of compositions containing small amounts of compounds being tested for reodorant properties by estimating the strength and quality of certain odours.

Much evaluation has been done using in vitno techniques for the screening of potential reodorant compounds. A procedure for carrying out such tests was designed for evaluation of such compounds and their masking effect on strong sources of odour, such as saliva which has been incubated by ageing in a test tube, onion, garlic and tobacco smoke. Incubated saliva exhibits the characteristic and typical malodour found in the mouth generally referred to as "bad breath" or halitosis.

Incubated saliva was prepared as follows: whole saliva was collected from random donors, pooled and filled into test tubes (10 cc./tube) and incubated at 37°C. for 18 hours. The tubes were cooled and then capned

Onion oil (commercially available) was prepared as an odour concentrate as follows: 2 cc. 95% ethanol; 2 cc. "Tween" 80; 0.1 cc. Onion Oil; q.s. water to 100 cc. of concentrate ("Tween" is a Registered Trade Mark). This stock solution was used to prepare dilute onion oil solutions containing 0.5 cc. stock + 249.5 cc. water.

A natural garlic oil concentrate stock solution was prepared as follows:

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2.0 cc. 95% ethanol 2.0 cc. Polysorbate 80 0.05 cc. garlic oil q.s. water to 200 cc.

This concentrate provided 5 cc. to 245 cc. 5 of water for use in test procedures.

Tobacco smoke odour was obtained by bubbling the smoke from 12 cigarettes through

400 cc. of 5% aqueous ethanol.

Two factors were considered to be of importance in defining a reordorant: the amount of a standard concentration of reodorant required to mask an odour and the volatility of the reodorant compound. The latter consideration is based on the possibility that the source of malodour is not destroyed so much as it is masked by the countervailing reodorant. The effectiveness of the reodorant is, therefore, partially dependent upon its ability to volatilise competitively with whatever malodour may be present.

The factors of effective concentration and volatility were determined for various possible reodorant compounds. Effective reodor-

ant concentration is determined as follows:

Five stoppered bottles are provided with 1 cc. of standard malodorous material, previously described, together with 1, 2, 3, 4 or 5 cc. of the reodorant compositions to be evaluated. A test panel of judges indicates which are still malodorous and to what degree and those which have no odour.

Tests which show an area around which the odour seems masked are further defined by adding reodorant to comparable 1 cc. samples in 0.2 cc. increments, starting at the last previous concentration at which malodour was still perceptible, to determine at what concentration the malodour is masked. This value is the number of cc. of reodorant solution required to overcome the specific malodour.

A second factor which is considered in screening potential reodorant compounds is that of volatility. Tests are made of compounds, in combinations with known sources of malodour, to determine whether, under as nearly equal conditions as possible, effective amounts of reodorant reach the nose, as compared to particular malodours. Judge make evaluations of malodour and reodorant combinations, noting the time elapsed in minutes and 1/100 ths thereof at which the malodour is judged to be masked by the reodorant compound. This test may proceed for a maximum of ten minutes.

The two numerical values arrived at in accordance with the above methods are multiplied. Compounds producing a reodorant value of 50 do not possess reodorant activity as defined herein.

How this reodorant value is arrived at is

shown in the following hypothetical table giving results of dilution of a source of malodour with a reodorant solution and the neutralising of a malodour with vapour from a reodorant:

	Dilution	Vapour Mixing*	Reodorant Value	
Compound A	1.6	5.0	8.0	70
Compound E	2.0	10.0	20.0	
Compound C		1.66	0.83	

*Time is given in minutes and hundreths of minutes. (Example: 1 minute 15 seconds becomes 1.25 minutes).

This concept of reodorant properties marks a new point of departure for the development of approaches to overcoming the problems of odour masking and at least aesthetically improving breath odour. It is also clear from the screening tests set forth below that the compounds of the present invention also possess qualities which suggests including them in aerosol room fresheners and other compositions designed to combat unwanted odours.

The reodorant value which has been selected as indicating compounds with no appreciable reodorant activity is 50; possible reodorant activity as deflected in a reodorant value sufficiently low to warrant further investigation would be about 30 and any value appreciably lower would be unquestionably active.

Once the reodorant concept was formulated, it appeared that it would be merely a matter of selecting obvious flavours and fragrances as probable reodorants. This did not prove to be the case. There was no consistency discernable between those compounds which were pleasant smelling of themselves and their effectiveness as reodorants. Several compounds tested not only possessed no reodorant effectiveness but, when used to mask unpleasant odours, produced a resultant odour worse than the original unpleasant odour.

An example of this occurred when incu- 105 bated saliva was mixed individually with cinnamic aldehyde and ambrettolide, both of which have pleasant fragrances but each of these in combination with incubated saliva resulted in a more objectionable odour than 110 the incubated saliva alone. Other compounds with slight to neutral fragrances unexpectedly displayed exceptional reodorant qualities on testing. Sixty-seven possibilities were screened, with the result that eleven were found to have 115 sufficient reodorant to warrant further experiment and the five compounds of the present invention have been shown to be superior reodorants.

A sampling of various terpenes is set forth 120 below. The reodorant values were arrived at in accordance with methods described above.

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		Reodo	orant Value		
	Reodorant	Saliva	Malodour Onion	Compound Garlic	Tobacco
5	Geranyl formate Geranyl acetate Geranyl propionate Geranyl butyrate	12.21 6.24 43.0 32.26	4.87 5.40 50.0 38.25	18.00 40.0 50.0	5.50 50.0
10	Geraniol Citral α - Ionone α - Methyl - Ionone	33.44 16.17 3.75 6.20	50.0 7.36 0.50	42.0 50.0 13.22 1.22	50.0 50.0 14.50 1.80
	β - Ionone	50.0			_

It is apparent, from a comparison of the results, that adjacent homologues do not have any necessary relationship with respect to reodorance.

We do not wish to be bound by any explanation of the phenomenon of reodorance but it may be theorised that this quality of a compound is due either to an ability to block odour receptor sites in the olfactory epithelium or to low olfactory thresholds for the compound and possibly a combination of both. This latter quality is the ease of detection of the reodorant compound when present in combination with other odours.

Further tests using human subjects having mouth odour problems were carried out using α - ionone as the reodorant. A test panel of five trained judges were used to monitor the breath of participants as to its odour. The participants are out of the judge's sight and their breath is conveyed to the judge's nose via a standardised cylindrical glass tube. A lapse of at least 90 seconds between samplings was adhered to, to prevent olfactory fatigue.

The *in vivo* methods had been refined by use of multiple judgings and replicate judgings. In the latter, a subject just rated was resubmitted to the judges for a second time, unknown to them, and the two evaluations compared. Judges scored the same or within 1 point of the previous score on a 1 to 9 rating scale about 85% of the time

scale about 85% of the time.

Three hundred and eleven subjects were tested using three flavour bases in conventional mouthwash bases which have included in them one or three flavorants, that of Flavour I, Orange Juice and Mint. These three formulations were used as such and with additions of α - ionone in amounts of 100, 500 and 1000 parts by weight per million.

A base value of breath odour was established using a random sampling of the test population using a scale of 1 to 9 with 5 indicating neutral or nearly odourless state while a lesser number indicates increasingly pleasant, while above 5 the higher the number indicates increasingly unpleasant odour. The standard for 9 was the odour of incubated saliva.

The results of the tests showed that the flavoured mouthwashes without reodorant reduced breath odour to different degrees with mint the most effective, Flavour I less effective and orange juice the least so. The improvement in breath odour quality was further increased when the mouthwashes were provided with increased amounts of α -ionone. This was the result when α - ionone was added in concentrations of 100, 500 and 1000 parts by weight per million.

The following Examples, which are given for the purpose of illustrating the present invention, represent embodiments which have proven to be particularly effective for obtaining the best results in breath odour improvement through the use of reodorant compounds:—

ounas: —			
Example 1. Glycerol USP 2. Sorbitol Solution US	100.0	g.	80
3. "Tween" 60 SD	6.0	g. g.	
4. SD Alcohol 38-F1)	182.7	ml.	
5. Sodium Cyclamate N	F		85
(optional)	1.6	g.	
6. Saccharin Sodium N	IF	Ü	
Powder	0.16	g.	
7. Flavour I ²⁾	0.753	g.	
8. FD & C Red #2 (10 dye)			90
9. FD & C Yellow #6	0.6	mg.	
dye)	10.0		
10. Sodium Phosphate	Dibasic	mg.	
Anhydrous	1.1	g.	95
11. Citric Acid Anhydro	ous USP.	8.	
Fine Granular	0.72	g.	
12. Water Purified		U	
USP	q.s. to 1000.0	ml.	
1) Alcohol SD 38F			100
Boric Acid, USP			100
Granulated	1.5100	æ	
Menthol USP	0.5526	g.	
0 . 0	0.5540	۵.	

0.0945 g.

ml.

105

180

Total Volume 182.7

Cassia Synthetic

Alcohol USP

						
	2) Flavour I Cassia	0.44176 g.	reodorant following order:	ed by others in	this descend-	
_	Orange Juice F-4521	0.31114 g.	Flavour I	—1000 parts preodorant	per million	
5	Lavandin Extra 30/32	0.00002 g.	Orange Juice Flavour	—1000 parts p	er million	65
	Orange Flavour Absolute	0.00002 g.	Flavour I	— 500 parts reodorant	per million	
	Method for Preparation.		Mint Flavour	- 500 parts reodorant		70
10	Step: A. Add 3 to 4 and mix.	Add 1 and	Orange Juice Flavour	— 500 parts reodorant		
	2. Continue mixing. B. Add 7 and 8 to A, mix for the continue B to 05% of finished	or 15 minutes.	Mint Flavour Orange Juice	-without reod - 100 parts		
15	C. Bring B to 95% of finishe 13. Add 5, 6, 11 and 12. A have dissolved and continu	Iix until solids	Flavour Flavour I	reodorant — 100 parts		75
	solution clears (approx. 45 D. Bring C to volume with	min.).	Mint Flavour	reodorant — 100 parts		
20	and allow to stand overnig temperature.	that ambient	Flavour I	reodorant —without reo	_	
20	E. Filter. F. Add 9 and 10 to E and	remiy	Orange Juice Flavour	-without reo		80
	The mouthwash prepared as		It is generall	y apparent that	the reodorant	
25 r	was then provided with α - reodorant in amounts of 100,	action of the compound increases with the increase in the amount present and that vari-			o #	
	parts per million by weight of composition. The resultant mouthwashes were judged to be		ous flavours in compositions also effect their odour-masking abilities.			85
	more effective at improving	the breath of panel of judges	in combination		gum, pressed	
30	using the organoleptic method	mint and candy lozenge formulations. The reodorant is usually supplied to these formulations in pure form and not in dilute solution.			90	
	mouthwash plus the various	amounts of ash above but	α - ionone is u	sed in its pure	form and not	
35	with the flavour replaced by n one set and orange juice flavo	nint flavour in ur in another,	an oily liquid	n, its pure form which is easily	compounded	95
	were evaluated. The combination evaluated v	vere:	All amounts given in parts 1	ingredients of in the following	examples are	
	1. Mouthwash as in Example	e I+100 ppm	given in parts i	Example II		
40	α - ionone. 2. Mouthwash as in Example	e I+500 ppm	Chewing Gur Gum base	n Î	20%	100
	 α - ionone. 3. Mouthwash as in Example 	I+1000 ppm	Sugar Corn Syru		64% 15%	
	α - ionone. Three compositions as in 1, 2	2 and 3 except	Flavour	Compounds	1% 10 —1 000	
45	that the flavour in each wa	s changed to		Example III	parts/million	105
	Three compositions as in 1, 2 that the flavour in each was cha	2 and 3 except	Pressed Mint Sugar		94.75%	
50	juice. Three compositions without		Corn Syru Magnesium		4% 1%	110
	using Flavour I, mint and ora	ange juice. these 12 com-	Flavour Reodorant	Compounds	0.25% 10 to 1000	
	binations of mouthwash bases v ant and with 100 parts per mil	lion, 500 parts			parts/million	
55	per million and 1000 parts per the Flavour of Example I, m	r million, with	Candy Lozer	Example IV Ige	(40)	115
	orange juice flavour were the	following:	Sugar Corn Syru	p	64% 35%	
	The combination having diminution of malodour was t	he mouthwash	Flavour Reodorant	Compounds	1% 10 to 1000	120
60	with mint flavour and 1000 pa	rts per million			parts/million	124

Reodorant compounds added within the E. Add 14 to 2/3 of final volume. Mix ranges indicated do not change the finished thoroughly. product physically in any essential manner. F. Add and dissolve 4, 5 and 6 in E. The reodorant compound's qualities are G. Q.S. to final volume with 14. Mix unimpaired by the processes used for the prethoroughly. 50 paration of these various compositions. H. Filter. Determine volume of filtrate. Example V J. Add 12 as 0.1% aqueous solution adjus-A preferred mouthwash composition is preted to filtrate volume (theoretically pared as follows: 210 ml./1000 ml.). 55 K. Add 13 as 1.0% aqueous solution adjus-10 1. Glycerol USP 50.0000 g. ted to filtrate volume (theoretically 2. Sorbitol Solution 1.0 ml./1000 ml.). USP 100.0000 g. L. Mix thoroughly. 3. Alcohol SD 38B3) for The resulting mouthwash was at least the 60 Reodorant Mouthwash equivalent to that of Example I, with mint 15 W6680-25 253.2000 ml. flavour and 1000 ppm α - ionone, with res-4. Sodium Saccharin NF pect to ameliorating breath odour. It has a Powder 1.2000 g. clear, green appearance and has the odour 5. Sodium Phosphate Dibasic and taste of spice mint. The taste is gener-0.1600 g. Anhydrous ally pleasing and lingers for some time after 20 6. Sodium Phosphate Monobasic Crystalline "Tween" 80 SD 1.2000 g. We disclaim any use of the present inven-15.0000 g. tion in the United Kingdom which is con-8. Menthol USP 0.0040 g. trary to the provisions of the Artificial 9. Imitation Mouth Refresher Sweeteners in Food Regulations 1967. 25 1.0000 g. (9/702559) 10. Mouthwash Flavour WHAT WE CLAIM IS:— V-30.278 1.0000 g. 1. A composition for the amelioration of 11. α - ionone 1.0000 g. malodours, comprising a known deodorant 12. FD & C Blue #1 (100% composition with a minor content of at least 75 30 dye basis) 0.0020 g. one compound selected from α - ionone, α -13. D & C Yellow #10 methyl - ionone, citral, geranyl formate and 0.0100 g. (100% dye basis) geranyl acetate. 14. Water, deienised 2. A composition according to claim 1, q.s. to 1.0000 L. wherein the reodorant is present in an amount, 80 by weight, of from 10 to 2000 parts per 35 3) Alcohol SD 38B million. 1. Menthol USP 1.5960 g. 3. Compositions according to claim 1 for 1.4000 g. 2. Peppermint Oil USP the amelioration of malodours, substantially 3. Alcohol 95% USP 250.0000 ml. as hereinbefore described and exemplified. 85 Method of Preparation. 40

A. Add 7 to 3 and mix well.

B. Add 8, 9, 10 and 11 and mix well. C. While mixing rapidly, slowly add 14 to approx. 1/2 of final volume. Mix until clear.

D. Add 1 and 2 to C and mix well. 45

VENNER, SHIPLEY & CO., Chartered Patent Agents, Rugby Chambers, 2, Rugby Street, London, W.C.1. Agents for the Applicants.

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